Soil solution fluxes and composition trends reveal risks of nitrate leaching from forest soils of Italy.
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#### 8 Abstract

9 Although acidification is a natural soil formation process, atmospheric emissions of nitrogen and 10 sulphur have exposed forest ecosystems to accelerated anthropogenic acidification for many decades. In Italy 11 nitrogen deposition loads are high, and among the highest in Europe. Data from the Italian ICP-Forests 12 monitoring network have shown that nitrogen deposition increases tree growth, and consequently the 13 organic carbon sequestration. However, the accumulation of nitrogen in the ecosystem could cause nutrient 14 imbalances and contribute to soil acidification. Within this context, we investigated the temporal changes of 15 atmospheric deposition and soil solution concentration in the Italian ICP-Forests using the non-parametric 16 Seasonal Mann Kendall (SMK) test. Further, we applied input-output models, to evaluate the capabilities of 17 the soil-forest system to retain deposited nitrogen and thus protect underground waters.

Increase of soil solution pH was observed in most of the sites, likely due to sulphate deposition 18 19 decrease with a concomitant high and stable natural exchangeable base deposition. For the sites around the 20 Po plain, however, high levels of nitrogen deposition impede pH increase. The site with the maximum mineral 21 N deposition showed signs of active acidification. The analysis of the fluxes of nitrogen compounds 22 demonstrates a complete retention of the ammonium forms, which further contribute to acidification 23 through nitrification. Furthermore, the Italian monitoring network showed that the retention of nitrogen in 24 the soil-forest system was effective since the observed N fluxes in mineral soil were strongly reduced compared to mineral N input in most of the plots. In spite of this, significant  $NO_3^-$  fluxes from the subsoil 25 26 were observed in sites with high deposition, and also in the southernmost site which is exposed to relatively

low mineral N loads. Drivers other than pollution should also be considered since N can follow different
pathways at different time scales, influencing N leaching independently from the amount of deposition.

29

30 1. Introduction

Since the late 1970s, post-industrial human activities caused a distortion of major biogeochemical cycles, arising from high emissions of sulphur (S) and nitrogen (N) from fossil fuel combustion and agriculture, and resulting in increased levels of acid deposition in many ecosystems (Phoenix et al., 2006; Galloway et al., 2008). Recently, several "planetary boundaries" (PB) have been proposed to define a safe operating space for humanity on a stable Earth system (Rockstrom et al., 2009). According to this concept, certain boundaries concerning biogeochemical fluxes have already been exceeded, especially those related to nitrogen cycles (Steffen et al., 2015).

High nitrate  $(NO_3^-)$  and sulphate  $(SO_4^{-2})$  concentrations in atmospheric deposition can be related to 38 water and soil acidification (Ronse et al., 1988; Norton et al., 2004; De Schrijver et al., 2006; Sherman et al., 39 40 2006; Stevens et al., 2009; Oulehle et al., 2010). Long-term excesses of N and S compound deposition cause 41 unfavourable conditions in forest soils (Verstraeten et al., 2012). Despite the positive function of N deposition 42 as a nutrient on forest growth rates, a potential negative effect at sites with high N deposition has been 43 recognized at continental scale in Europe (Etzold et al., 2020). Further, high nitrate concentrations in surface 44 and ground waters can lead to eutrophication of waters (Aber et al., 1998; Fenn et al., 1998; Rabalais, 2002) 45 and potential health problems (Briand et al., 2017). Critical limits for N concentrations in soil solution were 46 frequently exceeded between the early 1990s and 2006 in Europe (lost et al., 2012). With respect to the 47 acidifying effects of nitrogen deposition on soils, the potential contribution of ammonium (NH<sub>4</sub>) deposition 48 also needs to be considered, in addition to the effect of nitric acids, as microbial nitrification and assimilation 49 in soil involve the within-soil production of large amounts of protons (Rodhe et al., 2002).

50 In recent decades, emission abatement strategies, drawn within the Convention on Long-Range 51 Transboundary Air Pollution (CLRTAP; UNECE, 1979) in Europe, have resulted in large decreases in S emissions, while nitrogen emissions also decreased, but not to the same extent (EEA, 2018). In Europe,  $SO_4^{-2}$ and  $NO_3^-$  depositions decreased by 73% and 33%, respectively, between 1990 and 2012 (Colette et al., 2016). In Italy, emissions decreased by 93%, 63% and 19%, respectively for SO<sub>2</sub>, NOx and NH<sub>3</sub> from 1990 to 2016 (Taurino et al., 2018). As an effect of decreasing emissions, a substantial reduction in acid deposition has occurred in Italy (De Marco et al., 2019), as in most of Europe (Waldner et al., 2014; Karlsson et al., 2011).

A comprehensive study on the response of soil solution chemistry to decreasing acid deposition across Europe found that a reduction of acidifying atmospheric deposition has not produced clear signs of soil recovery (Johnson et al., 2018). The concomitant decrease of BCE at European scale may partly explain the observed low soil recovery (Johnson et al., 2018). Some authors have demonstrated how BCE deposition can also be due to industrial activities, such as mining (Davidson et al., 2020; Watmough et al., 2014) or quarrying (Oulehle et al., 2006) and that the general decrease of "heavy" industrial activities in the developed world is a likely cause of BCE deposition decrease.

64 Italy, due to its geographical position, receives large amounts of deposition from marine aerosol and 65 Saharan dust (Lequy et al., 2012; Costantini et al., 2018; Cecchini et al., 2019) and is subjected to a component of atmospheric deposition which is lesser or absent in northern European countries. The most represented 66 67 chemical species in this Mediterranean pattern of deposition are base cations and chloride, for which a 68 significant gradient of decrease from south to north is observed (Ehrmann et al., 2017; Cecchini et al, 2019). 69 Episodic inputs like those associated with Saharan dust events were registered in the Alps (Rogora et al., 70 2004) and can also have a significant impact on ecosystems in remote areas (Psenner, 1999). In the 71 meantime, anthropogenic nitrogen and sulphur depositions play a major role in Northern Italy. Some of the 72 highest N deposition loads are concentrated in the Po plain, one of the most urbanised and industrialised 73 areas of Europe. Studying the long-term trends of alkaline and acid compounds in both atmospheric 74 deposition and soil solutions is of interest for the Italian territory where sites cover different geographical 75 patterns. The soil solution composition provides an indication of nutrient availability (Sherman et al., 2006), 76 potential acid stress to vegetation (Oulehle et al., 2010; Binkley and Fisher, 2013), as well as on potential 77 exceedances of deposition critical loads. However, in contrast to studies on surface waters, the few regional Mediterranean studies on the response of soil solution chemistry to deposition (Avila et al., 1995, Rodà et al., 1990) focused on short time periods without capturing temporal trends. Long term assessments of changes in soil solution in relation to deposition are lacking for Mediterranean forest ecosystems, notwithstanding the peculiarities of the atmospheric deposition in this region.

82 Within this context, we aim to explore the specific impacts of peculiar atmospheric deposition, 83 hypothesising a strong influence of aeolian dusts and aerosol on forest soil ecosystem in Italy, as 84 representative of the Mediterranean region. Our main objectives are a) to evaluate the long-term trends in 85 soil solution concentrations in relation to deposition trends; b) to investigate the soil response to pollution changes, defining trends in soil solution pH; and c) to identify sites where the transfer of reactive N from 86 87 atmospheric pollution to groundwater is a potential concern. We tested the hypothesis that the processes of 88 Italian forest soil acidification, as indicated by soil solution composition, showed different trends compared 89 with those of other European countries. We expect a positive impact on soil solution pH due to a decrease 90 of atmospheric sulphate deposition whereas BCE deposition is constant. At the same time, we hypothesize 91 that the constant high levels of nitrogen deposition around the Po valley, a "hot N point" in the North of Italy, 92 could neutralize the positive impact of declining atmospheric sulphate deposition.

93 2. Materials and methods

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96 We used data of deposition and elemental concentrations in soil solutions recorded at six forest plots 97 of the ICP Forests Level II intensive monitoring network in Italy (Table 1 and Fig 1). The main site and soil 98 characteristics of LAZ1, EMI1, VEN1 and PIE1 were previously described by Cecchini et al. (2019). The other 99 two sites are both represented by beech forest (*Fagus sylvatica*). The ABR1 site is located at an altitude of 100 1500 m a.s.l. in the upper slopes of the calcareous Serra Lunga ridge, in the Apennines. "Terra Rossa" soils 101 develop on a hard, stratified, Cretaceous-age limestone and the soil parent-material is likely made up of a 102 mixture of limestone residue and aeolian dusts (Cecchini et al., 2002). A relevant contribution from air-fall

<sup>95 2.1</sup> Study sites

volcanic ashes is known for this area (Ferrari and Malesani 1973). The CAL1 area is located at an altitude of
 1100 m a.s.l., on the edge of the Aspromonte National Park (RC). It is rainy site, with an average annual
 precipitation of 1300 mm, and this balances for the relatively southern location of this beech forest.

106 2.2 Sample collection and analysis

For the 6 studied sites (Table 1), almost continuous data series covering the period 1998-2017 are available. Deposition was sampled using "bulk" samplers, i.e. funnels continuously exposed. Three bulk samplers were located in the open field, in a clearance close to the monitoring plot, while 16 samplers were located under forest canopy, to collect the "throughfall" deposition. Deposition sample collection and analysis were subject to quality assurance and quality control checks including the use of control charts for internal reference, charge balance and participation in laboratory ring tests (König et al., 2013).

113 The soil solution sampling started in 1997 with a single site (ABR1), was extended in 1999 to LAZ1 114 site and then progressively, to the other sites, as detailed in Table 1. Soil solutions were collected at several 115 depths. The gravity lysimeters (filtering plates) were inserted just below the forest floor (FF), except at the steeply sloping sites (PIE1). The tension lysimeters (suction cups) were inserted at three different depths 116 117 (15/20 cm, 40 cm and 60/70 cm), according to ICP-Forests (2011). The sampling took place bi-weekly, with 118 significant interruptions in winter, due to a permanent snow cover and freezing for the northern sites (PIE1 119 and VEN1). Sample treatment, preservation, analysis and quality control follow ICP-Forests (2011) and 120 preceding versions (Nieminen et al., 2016).

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2.3 Data handling and statistical analysis

For sulphur and nitrogen deposition, the monitoring results were compared with the results of the model developed by the EMEP Meteorological Synthesizing Centre – West (MSC-W), hosted by the Norwegian Meteorological Institute, for the modelling of transboundary fluxes of acidifying and eutrophying air pollutants (EMEP 2012). This model mainly consists in a source-receptor matrix estimating the contribution of the emissions to the depositions of main pollutants.

127 In this study, we analysed data from the forest floor (0 cm), representing the basic conditions and 128 trend of soil input, and from the subsoil (60 or 70 cm), which should be seen as a measure of the potential 129 risk of nitrate transfer to the watershed. Forest floor gravity lysimeters were characterized by high inter-130 annual variability and were not installed at all sites, thus data from the shallowest tension lysimeters (topsoil, 131 at 15 to 20 cm depth) were also systematically analysed. The soil mineral layer refers to the sum of topsoil 132 and subsoil. For each site at each depth, data from different collectors were averaged for all sampling periods.

Temporal changes were tested using the non-parametric Seasonal Mann Kendall (SMK) test for monthly atmospheric deposition and for monthly concentrations of soil solutions in R statistical environment (R version 3.5.0 (2018-04-23)) using rkt package (Marchetto, 2015). For the atmospheric deposition, the mean relative change per year (y<sup>-1</sup>) was estimated:

137 
$$rslope = \frac{b}{mean(y)}$$
 (1)

where *b* is the absolute long-term slope, and mean(y) is the mean value of the time series (Waldner et al., 2014). Results of the time series are then reported in % y<sup>-1</sup>. Ammonium in soil solution was close to or below the detection limit for most plots in the lysimeters installed in the mineral soil and only data from the organic layer were analysed for trends.

Spearman's rank correlation coefficient was applied as a nonparametric measure of rank correlation betweennitrogen deposition and precipitation.

## 144 2.3.1 Soil mineral N fluxes

145 The chloride tracer approach was adopted to estimate water flux through the soil as described by 146 Cecchini et al. (2019). Briefly, the water fluxes for each soil layer *I* were calculated as follows:

147 
$$Jwl = \frac{TFCl}{rCll}$$
 (2)

where *Jwl* is the water flux, in mm y<sup>-1</sup>, at layer *I*; TFCl is Cl<sup>-</sup>TF deposition, in mg m<sup>-2</sup> y<sup>-1</sup> and *rCll* is the median Cl<sup>-</sup> soil solution concentration at *I*, in mg l<sup>-1</sup>.

The flux of  $NO_3^-$  and  $NH_4^+$ , *Jil*, was estimated by multiplying the relevant *Jwl* value for the median soil solution concentration, *ril*, for the same layer and period. The overall budget for  $NO_3^-$  and  $NH_4^+$  (*Bi*) was then calculated as follows:

153 Bi = TDi - Jis (3)

- 154 where *s* is the layer of the deepest samplers.
- 155

156 3. Results

157 3.1 Long-term trends in deposition chemistry

158 Data on bulk and throughfall deposition are presented in Table 2 and in Fig. S1. All the time series 159 were long enough to guarantee the robustness of the results. No significant temporal changes were detected 160 for open field precipitation (BDH) and throughfall precipitation among all the studied sites. A seasonal Kendall test (SKT) performed on monthly  $SO_4^{-2}$  deposition values showed significant decreasing trend for all sites 161 162 (Table 3), with a marked year variability due to the high marine contribution to sulphate from the Mediterranean Sea. In deposition nitrogen was found in the form of the two major compounds: ammonium 163 164  $(NH_4^+)$  and nitrate  $(NO_3^-)$ . Year-to-year variability was less marked than in the case of sulphate deposition (Fig. S1), but a slight decreasing trend was shown by nitrate deposition (Table 3). For  $NH_4^+$ , significant 165 166 decreasing trends were found only in sites in Central Italy. BCE deposition was high (Table 2), stable, and no 167 significant changes were observed at all sites except for EMI1, where a decreasing trend was observed (Table

168 3).

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## 170 3.2 Soil solution chemistry and temporal trends

The sulphate concentration in soil solution (Fig. 2) showed change with depth for most of the sites, except for CAL1 and ABR1. For EMI1 and LAZ1 a significant increase of sulphate from the shallow to the deepest soil was observed. Nitrates were present in the organic layers at all sites, while differences in  $NO_3^-$ 

concentration were found in the mineral soil layers, and for CAL1, EMI1 and VEN1 the values were higher 174 than 1 mg N l<sup>-1</sup>, a proposed threshold for critical N leaching and saturation (Gundersen et al., 2006). A 175 common feature to all sites is the almost complete absence of  $NH_4^+$  in the mineral soils at both depths, 176 topsoil and subsoil. The base cation concentration was the lowest in the northern sites, and a latitudinal 177 178 trend was observed in the organic layers. Soil solution of the organic layers was moderately to slightly acidic 179 at most of the plots. The lowest levels of pH were found at the topsoil for EMI1, PIE1 and LAZ1 sites (Fig 2). 180 This pH increase with depth is in accordance with observations on soil solution chemistry at European scale 181 (Johnson et al., 2018). The observation of the time series of the forest floor (Fig. S2) clearly evidenced yearto-year variation. The  $SO_4^{-2}$  in the FF samplers showed a strong interannual fluctuation with different 182 behaviours between the monitoring sites (Fig. S2). No trend was found at the forest floor except for LAZ1 183 site, where  $SO_4^{-2}$  significantly decreased. At both analysed mineral layers, significant decreasing  $SO_4^{-2}$  trends 184 were dominant among the sites (Fig. 3). However, at the subsoil, for LAZ1 and for sites with historically high 185  $SO_4^{-2}$  deposition, as EMI1 and VEN1, no temporal changes were observed. Interestingly, these latter sites 186 were those that exhibited the largest decrease at the topsoil. 187

Nitrate concentration showed different trends between sites and depths.  $NO_3^-$  concentrations were 188 unchanged at all the depths at VEN1, while significantly increased at the deepest samplers for PIE1 and CAL1, 189 190 and at the topsoil for EMI1 (Fig. 3). In the forest floor,  $NO_3^-$  increased at EMI1 and LAZ1, and decreased at 191 ABR1 and CAL1. Trends for LAZ1 and ABR1 were detected only for the forest floor due to a sporadic presence of  $NO_3^-$  within the mineral soil that precluded any statistical processing. No significant changes in  $NH_4^+$ 192 193 concentration in soils solution at the forest floor were found for most of the sites. However, a decreasing trend was observed at LAZ1 (slope: - 0.03 mg  $l^{-1} y^{-1}$ , p<0.001) and CAL1 (slope: -0.02 mg  $l^{-1} y^{-1}$ , p<0.01) sites. 194 Increasing trends of the soil solution BCE concentration clearly dominated at most of the sites, except for 195 196 ABR1 and LAZ1, which had a decreasing trend in the forest floor and in the mineral layers, respectively. The 197 soil solution pH from the FF lysimeters showed no prevalent trends, except for VEN1 that showed a positive 198 trend (Fig 3).

#### 3.3 Seasonal patterns of N deposition

BD and TF deposition of  $NO_3^-$  and  $NH_4^+$  show little dependence on precipitation, as evidenced by the relatively low coefficients of correlation in the scatter plots at most sites (Fig. 4), except for EMI1 and PIE1. Monthly BD and TF deposition of  $NO_3^-$  and  $NH_4^+$  indicated an overall month to month variability, with different seasonal patterns between sites (Fig. S3). For the alpine sites (PIE1 and VEN1), a peak in spring and summer was observed for TF and BD deposition of both nitrogen compounds, while the other sites showed a spring maximum followed by a slow decrease of deposition in summer. At EMI1 a fall peak was observed for all kinds of deposition.

At most sites,  $NO_3^-$  and  $NH_4^+$  throughfall deposition were higher than bulk deposition (Fig. S3), with significant differences in the summer, from June to September. This is particularly evident for  $NO_3^$ deposition, while for  $NH_4^+$  this pattern was less marked at all sites, with very low differences between bulk and throughfall deposition throughout the year at PIE1 and for the sites with low N loads as ABR1 and LAZ1.

211 3.4 Nitrogen fluxes from the forest floor and subsoil

The outflow of N from the forest floor was higher than deposition at all the sites (Table 4), with a generally high contribution of  $NO_3^-$ . EMI1 also showed a high amount of  $NH_4^+$ . The principal seasonal pattern was the increase of the forest floor N outflow in spring (Fig. 5). The forest floor outflow decreased in summer due to scarcity of water flow, then raised to the maxima in fall, when the N accumulated in summer was leached (Fig. 5). At EMI1 high level of N outflow were observed also in summer. Differently from the other sites, VEN1 and ABR1 showed the maximum forest floor outflow of nitrogen in summer, when deposition was high and then regularly decreased to the end of the sampling period, despite increasing water flows.

N leaching from subsoil layers was completely inactive at LAZ1 and ABR1, limited at VEN1 and PIE1, and substantial at EMI1 and CAL1 (Table 4 and Fig. 6). The overall presence of  $NH_4^+$  was sporadic in the lysimeters at the mineral soil layers, thus mineral N leaching was normally made up of  $NO_3^-$ . At all sites, the amount of N leached out was much smaller than either forest floor outflow or throughfall deposition. Unfortunately, for the sites on the Alps, VEN1 and PIE1, our data on soil solution chemistry were available only for the period between May and November/December (Fig. 5 and 6); low temperatures did not allow to
 sample in winter. Thus, the incomplete time series provides only partial results on N leaching, and we can
 suppose that N leached is underestimated for PIE1 and VEN1.

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228 4. Discussion

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## 4.1 Impact of deposition on concentration and long-term trends of soil solution chemistry

230 The year-to-year high variability observed for soil solution concentration at the forest floor suggests a linkage with the climatic characteristics of the years that modulate processes at the forest floor- through 231 232 the indirect influence on forest stand and soil biological activity of microorganisms and pedofauna. These 233 processes likely masked changes due to atmospheric deposition. Thus, the discussion will be mainly focus on 234 the time series extracted from the mineral soil samplers, which show a much less complex behaviour with smoother long-term changes and limited seasonal effects. The deposition trends for  $SO_4^{-2}$ ,  $NO_3^{-}$  and  $NH_4^+$ 235 236 were within the range of relative slope observed for earlier studies on European deposition trends comparing 237 rslope values (Waldner et al., 2014). The most evident and generalized trend for all the sites was the concomitant decrease of  $SO_4^{-2}$  from atmospheric deposition and soil solution, thus confirming the 238 239 effectiveness of the abatement emission polices regarding S compounds. This result agrees with observations 240 at the European scale (Johnson et al., 2018) and from individual forest plots (i.e. Karlsson et al., 2011; 241 Vanguelova et al., 2010; Verstraeten et al., 2012; Waldner et al., 2014). At EMI1 and LAZ1 sites, however, a high  $SO_4^{-2}$  concentration and no long-term decrease were observed in soil solution at 60-80 cm. At the subsoil 242 243 lysimeters, Na<sup>+</sup> concentration was also high compared to the other sites (Cecchini et al., 2019). Past marine 244 intrusion might explain the high levels of Na<sup>+</sup> and  $SO_4^{-2}$ , compounds typical of marine environments. 245 Furthermore, for EMI1, despite a reduction of sulphur deposition in recent decades, the presence of sulphur-246 contaminated soils is plausible. Before the start of the monitoring, high acid loads may have caused sulphate 247 accumulation in soils around the Po plain (EMI1 and VEN1). For these sites, the absence of significant changes in soil solution of  $SO_4^{-2}$  at the subsoil implied a delay of response to observed decreasing sulphate deposition, 248

which may be ascribed to desorption of previously deposited  $SO_4^{-2}$  from soils (Fakhraei et al., 2016). This result is in agreement with higher release of  $SO_4^{-2}$  from soils found in sites with historically high deposition (Oulehle et al., 2006). For the most southerly sites,  $SO_4^{-2}$  peaks are likely linked to dry deposition of sulphateladen dusts or aerosols of marine and/or African origin, as was already found for ABR1 (Cecchini et al., 2002). Furthermore, the strong decrease of  $SO_4^{-2}$  at 20 cm (topsoil) might suggest that the desorption process is proceeding as a front from topsoil to subsoil, which is still saturated and is receiving large  $SO_4^{-2}$  concentrations from the topsoil, thus explaining high  $SO_4^{-2}$  concentration at the subsoil.

256 Inorganic N in soil solutions show no significant trends in response to decreasing nitrogen deposition in most of the sites. The decreasing trends of  $NO_3^-$  deposition seem too slight to act on mineral N in soil 257 solution (Table 3). Furthermore, a continuous contribution to N input due to  $NH_4^+$  deposition has to be 258 considered for all sites. The low percentage of plots with significant trends for NH<sub>4</sub><sup>+</sup> deposition, in the Italian 259 forests, is comparable with findings for the rest of Europe (Waldner et al., 2014). Agriculture is the main 260 261 source of NH<sub>3</sub> emissions (94% of the total; Taurino et al., 2018) and, although abatement technologies have 262 been introduced following the implementation of the EU IPPC Directive, NH<sub>3</sub> emissions are still too high in 263 Italy.

For EMI1, PIE1 and CAL1, an increasing trend in  $NO_3^-$  concentration has been recorded at the shallow and subsoil samplers, respectively. EMI1 and PIE1 sites were loaded by nitrogen ( $NO_3^- + NH_4^+$ ) throughfall deposition higher than 10 Kg N ha<sup>-1</sup> y<sup>-1</sup>, a threshold that corresponds to imbalance in input-output models (Dise and Wright 1995; Gundersen et al., 1998).

Aeolian dust contribution from Saharan sources that preferentially impact sites located in south (CAL1; Fig.1) and central Italy (LAZ1; Fig. S1; Rogora et al., 2004; Cecchini et al., 2019) likely also explain the highest level of BCE for these sites where large peaks were noticeable, since Sahara outbreaks were found to be conspicuous in atmospheric deposition particulate matter also at a site 1500 km away from the source (Lequy et al., 2018). The importance of the episodic deposition of Saharan dust was underscored by the large peaks in base cation deposition at the southernmost monitoring sites (CAL1; Fig. S1) and in Central Italy (LAZ1;

Fig. S1). This likely explains the observed unchanged trends for BCE at all sites. On the other hand, the significant negative trend observed at EMI, a site located in the Po Plain, can be explained by a concomitant decrease of the pollution component to BCE (Watmough et al., 2014).

277 At most of the sites, a positive pH trend was found for soil solution from mineral soil samplers. This 278 was likely related to a combining effect of a general decrease of sulphate and the increase of BCE in soil solution. These results could be interpreted as the most evident response to the decrease in  $SO_4^{-2}$  deposition 279 280 as was also suggested in previous studies (de Vries et al., 2003, Verstraeten et al., 2012; Fuss et al., 2015). On 281 the other hand, a negative topsoil pH trend was observed at EMI1, which is known to be a site with significant 282 exceedance of critical load for N deposition (Ferretti et al., 2014). This suggests that the effect of decreasing 283 S deposition can be neutralised by high N depositions. Furthermore, N at this site included a large amount of  $NH_4^+$ : the production of 2 moles of H<sup>+</sup> through nitrification of 1 mole of  $NH_4^+$  increases the strength of 284 285 acidification.

286 4.2 Nitrogen fluxes from the atmosphere to the subsoil.

The overall N throughfall deposition higher than bulk deposition suggests an important contribution 287 288 of dry deposition, which was previously recognised as the main form of atmospheric N input in 289 Mediterranean ecosystems (Sanz et al., 2002; Avila and Rodà 2012). The importance of dry deposition was 290 also recently supported by isotopic analysis conducted in Mediterranean holm oak forests (Guerrieri et al., 291 2019). In summer, total depositions generally decreased, but increased dry deposition causes TF depositions 292 to increase their contribution and, often, to reach the highest levels. The observed input pulses of nitrogen 293 into the forest soil after dry periods presumably originated from the washing of dry deposition accumulated 294 in the canopy (García-Gomez et al., 2016). The observed seasonal pattern in N TF deposition may also be 295 explained by a seasonal partitioning between atmospherically and biologically derived N-NO<sub>3</sub>; especially in 296 summer, after a drought period, a high proportion of NO<sub>3</sub><sup>-</sup>has been suggested to derive from nitrification at 297 the canopy (Guerrieri et al., 2019). At EM1 and LAZ1 sites, high spring N TF deposition, may also originate 298 from flowers and insects falling in the collectors and trapped on the filters.

299 From October to December, the most common pattern was for a general decrease; in the 300 southernmost site, CAL1, the decrease was marked with respect to precipitations. For most sites, deposition appeared to be controlled either by changes in source emissions or, more likely, by changes in atmospheric 301 302 circulation that either strengthen or weaken the connection with the source. The exceptions are EMI1 and 303 PIE1 sites, where deposition of both nitrogen forms was highest and strongly correlated with precipitation, 304 which was not observed for the other sites. It appears that sources of pollution are close, and emissions are 305 constant, so that transport is independent from atmospheric circulation. EMI1 and PIE1 are close to the "hot 306 N point" of the Po Plain, one of the most urbanized and cultivated area of Italy, and both NOx emission from 307 mobile sources and the largest source of NH<sub>3</sub> emissions due to livestock waste (Taurino et al., 2018) play a 308 key role, being a source of pollution also for forest sites.

309 At all sites, N released by the forest floor was higher than N deposition. This suggests that these soil-310 forest systems were well loaded with N, which was then intensely cycled. A general increase of flux in spring, 311 together with typical summer maxima at EMI1 and VEN1, were likely induced by increasing deposition at 312 most sites (Fig. S3) and by the influence of high temperature on mineralization rates within the forest floor. On the other hand, at the more southern sites (CAL1 and LAZ1), the low water availability in summer strongly 313 314 reduced the N outflow, while an increase was observed in fall. This was likely due to a release of nitrates 315 previously produced and accumulated in summer. Nitrification process was observed to begin at the canopy 316 level, on the leaf surface, by ammonia-oxidazing archea (Guerrieri et al., 2015) and in throughfall deposition 317 (Watanabe et al., 2016). Oxidizing microbes reach the soil via throughfall deposition and fallen leaves, further 318 explaining the maximum release of nitrate from the forest floor in fall for all sites (Fig. 5). The low flux of  $NH_4^+$  out from the forest floor compared with the  $NH_4^+$  throughfall input, and the high flux of nitrates leached 319 320 out of the FF (Table 4), likely suggest that ammonium was at least partly nitrified within the forest floor. The H<sup>+</sup> cation produced during nitrification possibly caused soil acidification. These observations confirm that the 321 high level of  $NH_4^+$  in atmospheric deposition represents a potential soil acidification factor in the northern 322 sites (EMI1, VEN1 and PIE1) where  $NH_4^+$  deposition were relatively high (Table 2). Furthermore, most soil N 323 324 stores in dead organic matter and can be converted from insoluble organic nitrogen to dissolved organic N

325 (DON) by microorganisms in natural forest ecosystems, and then converted in  $NH_4^+$  into the soil. Part of 326  $NH_4^+$  is assimilated by plants, immobilized by microbes, or absorbed by clay minerals.

A notable feature in mineral soil was the absence of  $NH_4^+$ , which represents a high proportion of 327 deposition and which is only partially retained in the forest floor (Table 4). At all sites, N was essentially 328 329 retained within the mineral soil. The organic layers of most of the Italian forests is normally too thin to 330 function as a storage (Andreetta et al., 2011), although it seems to be very active, being more a nitrogen source than a sink. The absence of  $NH_4^+$  in soil solution at the deepest samplers is in line with what was 331 332 observed for other European countries (eg., MacDonald et al., 2002). This is attributed to the nitrification 333 process, converting  $NH_4^+$  into  $NO_3^-$  (Gilliam et al., 2001) after possible temporary adsorption on soil negative charges, thus impeding fast leaching. 334

Although the seasonal distribution showed different patterns among sites (Fig. 6), N leaching 335 336 mostly took place in fall and spring. The peaks observed during the vegetative season followed either high 337 deposition (Fig S3) or FF outflow (Fig. 5). The winter peak at CAL1 appears to be associated with low 338 biological uptake and high water flow. The different characteristics of the soils are likely to play a role. Soils 339 at EMI1 and VEN1 are fine-textured, with long residence times for water; this maximizes uptake 340 opportunities when biological activity is high, thus explaining low level of N leached in spring despite high 341 water flow. On the contrary, soils in CAL1 and PIE1 have high permeability and short water residence times, 342 so that peak N inputs at forest floor/mineral soil interface are transferred downwards.

Fluxes of nitrogen through the studied forest-soil ecosystems thus appear to be rather complex. N deposition indirectly induced a state of N saturation that caused a high flux of N from the forest floor. In turn, N flux was heavily controlled by the effects of the mineralization rate and the water flow, thus following time patterns independent from deposition. High N forest floor outflow is very efficiently retained within the mineral soil. Leaching of N at subsoil level is absent or small in most sites (Table 4) and is important at the very-high-deposition EMI1 site but also at CAL1, despite its relatively low N deposition loads (Table 4). At EMI relevant deposition caused a continuous nitrate loss from the soil, but the observed annual outflux is lower

than at CAL1 (Table 4). This might be explained by lower precipitation and higher soil clay content at EMI1, with likely higher accumulation of N in mineral soils (Fuss et al., 2019). Moreover, microbial N uptake could lead to a significant long-term sink for N in forests if that N is subsequently transferred to stable soil organic matter (Curtis et al., 2011). Differences in SOM mineral association may influence soil biogeochemical cycling at different timescales, leading to different response to atmospheric deposition.

Furthermore, the forest at CAL1 is reaching mature age and trees are showing signs of distress in the last few years; consequently, it would be expected to have declining N accumulation in biomass. This is in agreement with conceptual models of nutrient retention in ecosystems for which mature forests should experience increased nitrate ( $NO_3^-$ ) losses in response to decreased plant demand (Aber et al., 2002; Pourmokhtarian et al., 2012). On the other hand, the EMI1 forest is undergoing succession (Bertini et al., 2011), thus vegetation is growing rapidly and nitrogen (N) leaching to groundwater should be lower than expected.

362 These results on N leaching in Italian forest soils demonstrate that data on atmospheric deposition 363 alone does not allow us to always predict the nitrogen fate in the soil. The complex nature of plant-soil 364 feedbacks involved in the nitrogen cycle has to be considered, since the interchange of N between vegetation 365 and soil can decouple N export and biomass dynamics, and cause forest ecosystems to retain N longer than 366 expected (Lovett et al., 2018). Further, N dynamics depend on N-mineral association (Fuss et al., 2019) and 367 on the nature of soil organic matter that emergent views describe as a continuum of progressively 368 decomposing organic compounds (Lehmann and Kleber 2015). Thus, N can follow different pathways at 369 different time scales influencing N leaching independently from the amount of deposition.

370

371 5 Conclusions

The evaluation of the time series for atmospheric deposition and soil solution chemistry provides evidence of reductions in  $SO_4^{-2}$  deposition and soil water  $SO_4^{-2}$  concentrations at all Level II intensive monitoring sites across Italy, confirming the successful implementation of the emission reduction policies.

375 However, concerns over the local effects of excess nitrogen deposition pollution remain. Despite slightly decreasing  $NO_3^-$  deposition, high  $NH_4^+$  deposition to northern Italian forests still appears to continue at a 376 constant rate. For the site (EMI1) located in the Po plain, a "hot spot" concerning N deposition, soil solution 377 378 pH is decreasing at the topsoil. For all the other sites, a positive response of soil solution pH to decreases in 379 acid deposition is evident. Although acidification does not seem to be a real concern for most of the forest 380 soils in Italy, the analysis of nitrogen fluxes from the atmosphere to the subsoil points to a potential risk of 381 nitrate leaching. For inorganic N, the decreasing trends seem too slight to avoid exceedance of critical loads 382 in the north, where further reduction of N emissions is needed to prevent air pollution effects on forests. 383 Furthermore, factors other than pollution have to be considered, as evidenced by CAL1, a site in the south of 384 Italy where, despite the relative low N atmospheric deposition, nitrate leached out from the subsoil is 385 significant. The large variability of the Italian territory in terms of climate, geology and soil type requires 386 careful monitoring of the forest ecosystems in order to follow up chemistry trends and fluxes in soil solutions 387 with particular attention to nitrogen leaching concerns.

388

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# 610 Figure captions

- 611 Fig. 1 Location of the 6 studied sites.
- Fig. 2 Concentration of S- SO\_4^(-2), N-NO\_3^-, base cations and pH for the forest-floor (0 cm), topsoil (20 cm) and subsoil (60 cm) samples at each site.
- Fig. 3 Results of the trend analysis by Seasonal Kendall Test (SKT) applied to monthly data of sulphate, nitrate,
- ammonia and base cation concentrations at the forest floor (0 cm), topsoil (20 cm) and subsoil (60 cm) for each site. Significance of Sen's slope: \* $p \le 0.05$ ; \*\* $p \le 0.01$ ; \*\*\* $p \le 0.001$ ; n.s.: not significant.
- Fig. 4 Scatterplot of precipitation (mm) and deposition of N-NO\_3^- (red) and N-NH\_4^+(black). Numbers at
  the top-left indicate the Spearman's coefficient of correlation.
- Fig. 5 Representation of the monthly fluxes of water (Jw -bars) and nitrates (JNO3-lines) out of the forest floor for each site. Dashed lines represent the minimum and the maximum recorded nitrate flux within each time series.
- Fig. 6 Representation of the monthly fluxes of water (Jw -bars) and nitrates (JNO3 -lines) out of the deepest layer for each site. At ABR1 and LAZ1 sites, no presence of nitrates was observed in the deepest samplers.
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642 Fig. 4



Fig. 5



648 Fig. 6

651 Table 1

652	Main characteristics	of the	sampling	sites	and s	soils.
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Code	Positio	n Altitude	Precipitation	Vegetation	Soil	BS*	Period for
		(m a.s.l.)	(mm)			(%)	trend
							analysis**
CAL1	38°25'N, 16°10'I	E 975	1880	Fagus sylvatica	Haplic Umbrisol	13	2009-2017
ABR1	41°51'N, 13°34'I	E 1500	950	Fagus sylvatica	Humic Alisol	75	1997-2017
LAZ1	42°49'N, 11°54'I	E 675	1019	Quercus cerris	Cutanic Alisol	48	1999-2017
EMI1	44°43'N, 10°12'I	E 225	859	Quercus petraea, Q. cerris	Endostagnic Luvisol	51	2010-2017
PIE1	45°41'N, 8°04'I	E 1150	1811	~ Fagus sylvatica	Haplic Cambisol	16	2006-2017
VEN1	46°03'N, 12°01'I	E 1125	1900	Fagus sylvatica	Haplic Luvisol	100	2010-2017

653 \*Soil base saturation (B horizon)

654 \*\* Soil solution655

## 657 Table 2

Average annual deposition for the time series (1998 to 2017) among sites

	A					
Site	Bul	k deposition	n (kg N ha <sup>-1</sup>	y <sup>-1</sup> )	Throughfall deposition (H	kg N ha <sup>-1</sup> y <sup>-1</sup> )
	$N-NO_3^-$	N-NH <sup>+</sup>	$S-SO_{4}^{2-}$	BCE	N-NO <sub>3</sub> N-NH <sub>4</sub> S-S	$0_4^{2-}$ BCE
CAL1	3.47	2.85	34.96	67.28	4.01 3.23 53	3.09 133.18
ABR1	3.31	2.68	12.99	29.90	3.48 2.10 12	2.45 47.10
LAZ1	3.32	2.42	14.70	27.20	4.56 2.07 18	8.10 54.17
EMI1	4.84	6.68	12.87	12.31	8.15 10.34 18	3.76 37.50
PIE1	7.34	7.97	20.18	13.46	7.99 7.68 19	9.16 23.92
VEN1	5.01	12.73	20.16	22.37	5.81 6.80 18	8.91 42.96

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# 661 Table 3

662 Results of the trend analysis by Seasonal Kendall Test (SKT) applied to monthly data of atmospheric

deposition, expressed as relative change for the period 1998-2017.

Site	Throughfall deposition (% y <sup>-1</sup> )												
	$N-NO_3^-$	N-NH <sub>4</sub> <sup>+</sup>	BCE	S-S0 <sub>4</sub> <sup>2-</sup>									
CAL1	n.s.	n.s.	n.s.	-4.03	*								
ABR1	-7.32	* -4.97 *	n.s.	-4.37	*								
LAZ1	-3.2	* -1.38 *	n.s.	-7.27	***								
EMI1	-2.41	* n.s.	-2.1	* -10.13	***								
PIE1	-4.08	*** n.s.	n.s.	-5.96	***								
VEN1	-3.78	** n.s.	n.s.	-5.29	**								

664 Level of significance: \*\*\*p < 0.001; \*\*p < 0.01; \*p < 0.05

<sup>656</sup> 

	CAL1		CAL1 ABR1		LAZ1		EMI1		PIE1 <sup>§</sup>			VEN1 <sup>§</sup>						
	N-NO <sub>3</sub> N-NH <sub>4</sub> <sup>+</sup> N <sub>tot</sub>		N-N0 <sub>3</sub>	$N-NH_4^+$	$N_{\text{tot}}$	N-N0 <sub>3</sub>	N-NH <sub>4</sub> <sup>+</sup>	N <sub>tot</sub>	$N-NO_3^-$	N-NH <sub>4</sub> <sup>+</sup>	Nt <sub>ot</sub>	N-N0 <sub>3</sub> N	I-NH <sub>4</sub> <sup>+</sup>	$N_{\text{tot}}$	N-N0 <sub>3</sub>	N-NH <sub>4</sub> <sup>+</sup>	$N_{\text{tot}}$	
	(kg N ha <sup>-1</sup> y <sup>-1</sup> )		(kg N	ha <sup>-1</sup> y <sup>-1</sup>	)	(kg N ha <sup>-1</sup> y <sup>-1</sup> )		(kg N ha <sup>-1</sup> y <sup>-1</sup> )		(kg N ha <sup>-1</sup> y <sup>-1</sup> )		(kg N ha <sup>-1</sup> y <sup>-1</sup> )						
BD	2.8	3.2	6	3.5	3.1	6.6	3.6	2.6	6.2	4.9	6.3	11.2	2.9	3.5	6.4	2.4	8.5	10.9
TFD	4.5	4.6	9.1	4.4	2.7	7.1	6.4	1.9	8.3	7.7	10.4	18.1	3.6	3.7	7.3	3.7	4.5	8.2
$\mathbf{J}_{\mathrm{ff}}$	8.4	0.8	9.2	5	0.2	5.2	8.4	2.3	10.7	20.9	5.2	26.1	-	-	-	15.8	0.3	16.1
$\mathbf{B}_{\mathrm{ff}}$	-3.9	3.8	-0.1	-0.6	2.5	1.9	-2	-0.4	-2.4	-13.2	5.2	-8	-	-	-	-12.1	4.2	-7.9
$\mathbf{J}_{\mathbf{s}}$	3.3	0	3.3	0	0	0	0	0	0	2.6	0	2.6	0.8	0	0.8	0.9	0	0.9
B <sub>TFD</sub>	1.2	4.6	5.8	4.4	2.7	7.1	6.4	1.9	8.3	5.2	10.4	15.6	2.8	3.7	6.5	2.8	4.5	7.3

666 Components of the  $N_{min}$  soil budget (TF-based estimate, in kg ha<sup>-1</sup> y<sup>-1</sup>): J<sub>ff</sub>: flux out of the forest floor; B<sub>ff</sub>: budget of the forest floor; J<sub>s</sub>: flux out of the deepest 667 layer; B<sub>TFD</sub>: budget with TFD as input (from starting soil solution sampling year to 2015).

<sup>§</sup>For PIE1 and VEN1 sites, since soil solution sampling cover only the period from May to November, also the data reported for bulk and throughfall deposition are related to
 this period. The values of N deposition for the full year are reported in Table 2.