



Article Dihydroxyphenyl- and Heteroaromatic-Based Thienopyrimidinones to Tackle HIV-1 LEDGF/p75-Dependent IN Activity

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Abstract: The spread of Human Immunodeficiency Virus (HIV) still represents a global public health issue of major concern, and would benefit from unveiling unique viral features as targets for drug design. In this respect, HIV-1 integrase (IN), due to the absence of homologs in human cells, is a popular target for the synthesis of novel selective compounds. Moreover, as drug-resistant viral strains are rapidly evolving, the development of novel allosteric inhibitors is acutely required. Recently, we have observed that Kuwanon-L, quinazolinones and thienopyrimidinones containing at least one polyphenol unit, effectively inhibited HIV-1 IN activity. Thus, in the present research, novel dihydroxyphenyl-based thienopyrimidinone derivatives were investigated for their LEDGF/p75-dependent IN inhibitory activity. Our findings indicated a close correlation between the position of the OH group on the phenyl moiety and IN inhibitory activity of these compounds. As catechol may be involved in cytotoxicity, its replacement by other aromatic scaffolds was also exploited. As a result, compounds **21–23**, **25** and **26** with enhanced IN inhibitory activity provided good lead candidates, with **25** being the most selective for IN. Lastly, UV spectrometric experiments suggested a plausible allosteric mode of action, as none of the thienopirimidinones showed Mg²⁺ chelation properties otherwise typical of IN strand transfer inhibitors (INSTIs).

Keywords: HIV-1 integrase (IN); allosteric inhibitors; thienopyrimidinones; LEDGF/p75-dependent IN activity

1. Introduction

According to the World Health Organization, HIV transmission still represents an important worldwide public health problem, with around 1.3 million new infections worldwide by 2022 [1] mainly related to HIV-1, the agent responsible for the acquired immunodeficiency syndrome (AIDS).

Although many therapeutic strategies have been developed to strengthen the cART arsenal, the evolution of drug-resistant viral strains remains a serious issue. Thus, identifying novel viral factors as targets for drug design remains a priority. In this regard, HIV-1 integrase (IN) is an attractive target for new systematic drug discovery efforts, especially because of the absence of a cellular homolog that may result in a reduced occurrence of adverse effects [2].

IN is a multimeric enzyme and, together with reverse transcriptase (RT) and protease, is one of the three viral enzymes necessary for HIV-1 replication. It catalyzes the incorporation of double-stranded proviral DNA into the host cell genome [3] and comprises three domains: the N-terminal region, the central core domain containing the catalytic triad DDE



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Copyright: © 2023 by the authors. Licensee MDPI, Basel, Switzerland. This article is an open access article distributed under the terms and conditions of the Creative Commons Attribution (CC BY) license (https:// creativecommons.org/licenses/by/ 4.0/). that chelates two Mg²⁺ ions, and the C-terminal region [4]. In particular, the integration process requires the following two steps: (1) in the cytosol, the IN dimer catalyzes cleavage of a guanine–thymine (GT) dinucleotide from the 3'-terminus of each viral DNA end (3'-*processing*) and (2) in the nucleus, tetrameric IN mediates integration of recessed viral DNA ends into the cellular DNA via a concerted hydroxy-mediated nucleophilic reaction (*DNA strand transfer*).

The advent of a convenient in vitro functional assay based on recombinant IN [5] prompted drug discovery efforts to inhibit its DNA strand transfer activity by targeting the active site. As a result, strand transfer inhibitors (INSTIs) Raltegravir (RAL) [6,7], Elvitegravir (EVG) [8], Dolutegravir (DTG) [9], Cabotegravir [10] and Bictegravir [11] have been developed and approved for clinical use. Regrettably, the rapid emergence of viral variants displaying cross-resistance to the FDA-licensed INSTIs highlights the need to develop inhibitors with unique modes of action.

In this regard, the identification [12] and confirmation [13–18] of Lens Epithelium-Derived Growth Factor (LEDGF/p75) as an HIV-1 IN binding partner has serendipitously opened up a new route in antiviral drug design, and a series of novel allosteric inhibitors of the LEDGF/p75 binding site (LEDGINs) preventing viral replication by inducing inactive or aberrant IN oligomerization have been reported [19–24]. However, the development of small molecules (*allosteric inhibitors*) that access IN elsewhere, and which dramatically hamper activity while still retaining potency against INSTIs and LEDGIN-resistant viral strains, is urgently warranted.

Interestingly, a recent high throughput screening study provided a series of different promising hit to lead IN-inhibitor scaffolds from which the (1-(*tert*-butoxycarbonyl)-4-phenylpiperidine-4-carbonyl)phenylalanine 4 was selected and used for further optimization [25] (Figure 1). In addition, the inhibitory activity of various natural flavonols containing polyphenol moieties, e.g., angoletin, quercetin, myricetin and kaempferol has been previously reported [26–28].

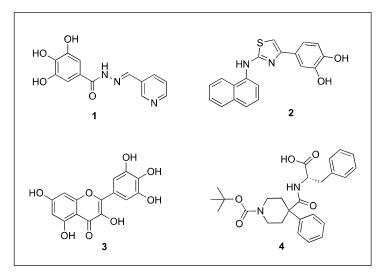


Figure 1. Examples of chemical scaffolds obtained from high throughput screening approach.

The increasing interest in novel allosteric IN binding sites prompted some of us to investigate the recently discovered sucrose binding pocket, identifying Kuwanon-L (Figure 2) as a novel allosteric inhibitor. In silico investigation supported the biological findings, also shedding light on the possible binding mode of the polyphenolic scaffold within the sucrose binding pocket [29].

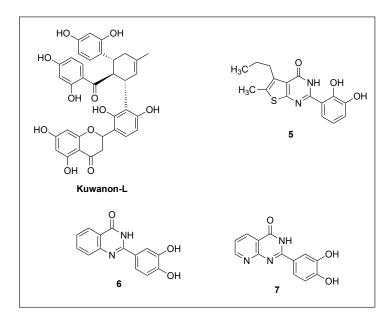


Figure 2. Chemical structure of Kuwanon-L and of most active thienopyrimidinones and quinazolinones.

Recently, we also reported on the ability of some 3,4-catechol-thienopyrimidinone [30,31] and quinazolinone derivatives [32], originally designed to allosterically inhibit the HIV-1 RT-associated Ribonuclease H (RNase H) function, to hamper LEDGF/p75-dependent IN activity. Interestingly, with regard to thienopyrimidinones, we observed that the 2-(2,3-dihydroxyphenyl)-6-methyl-5-propylthieno[2,3-d]pyrimidin-4(3H)-one (5), bearing a 2,3-catecol moiety, displayed the best IN inhibitory activity [30]. Figure 2 shows the most active compounds.

The thienopyrimidinone framework is widely documented in many biologically active compounds such as anti-tumor [33], anti-malarial [34,35], anti-microbial and antifungal [36], anti-inflammatory [37] and anti-tubercular [38] agents.

Thus, in the present paper, we investigated the LEDGF/p75-dependent IN activity of a series of dihydroxyphenyl-thienopyrimidinones using the strand transfer inhibitor Raltegravir as a positive control. Our results demonstrated a strict relationship between the position of the OH groups on the phenyl moiety and their IN inhibitory activity. Moreover, UV spectrometric experiments demonstrated that all of the thienopirimidinones were unable to chelate Mg^{2+} , thereby suggesting a possible allosteric mode of action. As the catechol scaffold—although present in many biologically active compounds—may be associated with cytotoxicity [39,40], we also synthesized and explored the inhibitory activity of differently substituted thienopyrimidinones. They were also examined for their RT-RNase H activity using 2-(3,4-dihydroxyphenyl)-5,6-dimethylthieno[2,3-d]pyrimidin-4(3H)-one (GZ 510) [30] as a positive control.

2. Results

Methods of investigation are shown in Figure 3.

2.1. Chemistry

Figure 4 presents thienopyrimidinones investigated in the present research.

As reported in Scheme 1, the novel thienopyrimidinones were synthesized according to our previously reported procedure for preparation of compounds **8**, **9**, **13**, **14**, **15–18**, **23** and **26** [30]. In brief, a Gewald reaction between cyanoacetamide, the appropriate carbonyl derivative, and elemental sulfur in the presence of diethylamine (DEA) as a base, was firstly carried out to afford the correspondent 2-amino-4,5-disubstituted-thiophene-3-carboxamide. Subsequently, the thiophene intermediate reacted via direct oxidative condensation with the proper aldehyde in the presence of molecular iodine, to give rise to the final products **10–12**, **19–22**, **24**, **25**, **27** and **28**.

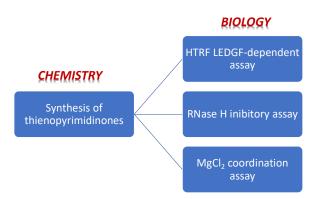


Figure 3. Methodology flowchart.

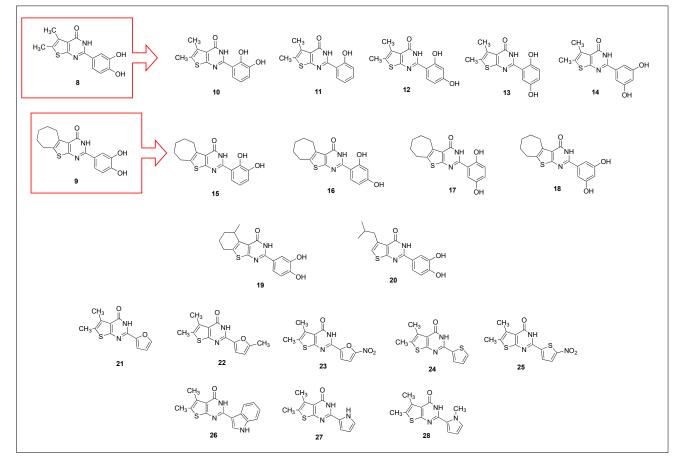
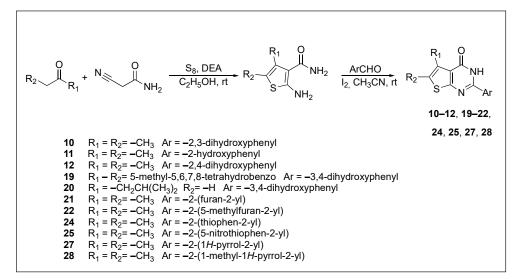


Figure 4. Chemical structures of thienopyrimidinones tested for their effect on the IN LEDGF/p75dependent activity.

2.2. Biology

All of the thienopyrimidinones were tested for their ability to affect the IN-LEDGF/p75 dependent activity.

From the analysis of the first series of compounds, it was immediately evident how the relative OH position on the aromatic core influenced the inhibitors' IC_{50} (Table 1). Remarkably, moving the OH from 4- to 2-position seemed determinant for the improved activity of compound **10**. Other possible combinations were explored, but the distancing of the second OH group negatively affected the inhibitory activity of compounds **12** and **13**. The 3,5-dihydroxyphenyl core appeared to be the best combination, and 2-(3,5-dihydroxyphenyl)-5,6-dimethylthieno[2,3-d]pyrimidin-4(3H)-one **14** displayed the best IC_{50} value.



Scheme 1. Synthetic pathway for compounds 10–12, 19–22, 24, 25, 27 and 28.

Table 1. Effect of thienopyrimidinones 8, 10–14 on the HIV-1 IN LEDGF/p75-dependent activity.

Compound	CORE	IC ₅₀ (μM)
8 [30]	ОН	24.4 ± 2.60
10	OH	12.6 ± 1.29
11	OH	21.96 ± 2.8
12	ОН	22.26 ± 1.57
13	ОН	85 ± 6.00
14	OH	3.7 ± 0.73
RAL	ÓH	0.055 ± 0.002

Introducing more bulky substituents on the thiophene scaffold showed a different trend. In this case, distancing of the second OH group from the 2-hydroxy moiety improved inhibition of the IN LEDGF/p75-dependent activity, as demonstrated by the 3,5,6,7,8,9-hexahydro-4*H*-cyclohepta[4,5]thieno[2,3-*d*]pyrimidin-4-ones **16** and **17**. Once again, the 3,5-dihydroxyphenyl unit proved the optimal core, with compound **18** being the best inhibitor of both series of thienopyrimidinones (Table 2 and Figure 5).

Compound	CORE	IC ₅₀ (μM)
9 [30]	OH	18.3 ± 2.3
15	OH	24 ± 1
16	ОН	10 ± 1.00
17	ОН	5.65 ± 0.15
18	OH	3.37 ± 0.48
RAL	5	0.055 ± 0.002

Table 2. Effect of thienopyrimidinones 9, 15–18 on the HIV-1 IN LEDGF/p75-dependent activity.

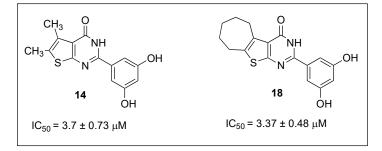


Figure 5. Chemical structure of dihydroxyphenyl thienopyrimidinones with the best IN inhibitory activity.

These results prompted us to explore substitution on the thiophene ring, and the most interesting compounds are reported in Table 3. Replacing the cyclohepthyl ring of **9** with a branched cyclohexane dramatically improved the inhibitory activity of compound **19** (Figure 6). Similarly, substitution of methyl groups of **8** with the bulkier and more flexible isobutyl moiety generated the more active thienopyrimidinone **20** (Table 3).

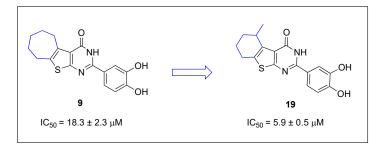


Figure 6. Effect of the replacement of cyclohepthyl ring with a branched cyclohexane on IN inhibitory activity.

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Compound	STRUCTURE	IC ₅₀ (μM)	
19	NH S N OH	5.9 ± 0.5	
20	O NH S N OH OH	8.5 ± 2.5	
RAL		0.055 ± 0.002	

Table 3. Effect of the thiophene ring substitution on the HIV-1 IN LEDGF/p75-dependent activity.

As the presence of dihydroxyphenyls may induce cytotoxicity [39,40], we addressed the possibility of replacing the dihydroxyphenyl moiety with other aromatic scaffolds, testing the inhibitory activity on both IN and RNase H functions (Table 4). Compounds 21–23 and 26 inhibited both IN and RNase-H activity, with 22 and 23 being more selective against IN, while introducing a -NO2 group on the thiophene core of 24 totally reversed its activity, making compound 25 completely IN selective (Table 4 and Figure 7). Intriguingly, both pyrrole compounds 27 and 28 were completely IN inactive, while the indole derivative **26** showed a good activity on both IN and RNase-H function (Table 4 and Figure 7).

Table 4. Effect of thienopyrimidinones 21-28 on the HIV-1 IN LEDGF/p75-dependent activity and RT-RNase H.

Compound	CORE	IC ₅₀	(μM)
		IN	RNase-H
21	\o>	5.75 ± 0.85	2.5 ± 0.1
22	CH3	8.75 ± 0.35	14.4 ± 0.2
23		4.6 ± 0.82	16.9 ± 1.7 [30]
24	∠ s	>50	8.2 ± 1.3
25	S NO ₂	3.2 ± 0.80	>50
26	N H	2.85 ± 0.85	$0.48\pm0.6~[30]$
27	₩ N CII	>50	35.3 ± 1.5
28	⊂H ₃ N	>50	14.8 ± 0.6
RAL GZ 510 [30]		0.055 ± 0.002 –	0.26 ± 0.01

0 CH₃

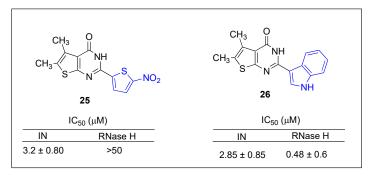


Figure 7. Chemical structure of the most promising heteroaromatic-based thienopyrimidinones.

Finally, to investigate the possible IN inhibition mechanism, active thienopyrimidinones were tested for their metal-chelating properties regarding Mg²⁺ ions implicated as cofactors in the mode of action of INSTIS [41]. Interestingly, although catechol-based compounds are especially notorious for their metal complexing properties, our UV spectrometric experiments excluded the participation of Mg²⁺ cations in their mode of action (see Supplementary Material).

3. Discussion

A series of thienopyrimidinones have been tested for their in vitro inhibitory activity on IN catalytic function. We chose the LEDGF/p75-dependent integration assay, as it is a versatile tool for detecting alteration of various IN functions such as multimerization, LEDGF/p75 binding and catalytic activity [25].

Structurally, a 3,4-catechol unit at the C-2 position of the thienopyrimidinone scaffold, regardless of any other substitution on the thiophene core, was a key factor in consistently fostering HIV-1 RT-RNase H but not IN activity [30]. As a matter of fact, in our first experiment we observed that switching from 2,3- to 3,4- catechol turned **5** in the IN inactive compound **29** (IC₅₀ > 50 mM) (Figure 8).

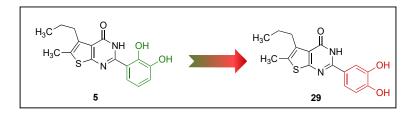


Figure 8. Chemical structure of compounds 5 and 29.

These observations prompted us to further investigate the relationship between the position of OH substitutions and IN LEDGF/p75-dependent activity.

Firstly, compounds 8 and 9 were selected as a baseline, since they were the most active against RNAse H and very poorly active against IN (Figure 9), and several derivatives exploring all the possible OH combinations were investigated.

The results shown in Tables 1 and 2 are suggestive of a cooperative relationship between the OH relative position and the thiophene substitution. In particular, as also evidenced in Table 3, bulkier and branched alkyl substitution in 5,6 positions may enhance IN inhibitory activity.

Moreover, compound **11**, bearing only a 2-hydrophenyl core, was slightly more active than **8**, hypothesizing a cooperative intramolecular hydrogen bonding between amide and hydroxy moiety.

Although the catechol unit is present in a wide variety of biologically active compounds, there is no doubt that its redox reactivity may be related to cytotoxicity [39,40]. Thus, a series of thienopyrimidinones supporting different heteroaromatic cores have been investigated (Table 4). Our first candidate, the electron-rich furan ring often used as a bioisosteres of benzene core [42] provided a valid alternative, leading to compound 21 (RNase H IC₅₀ of 2.5 ± 0.1 and IN IC₅₀ of 5.75 ± 0.85). Introducing an electron-donating methyl group in 5-position resulted in a reduced activity of 22, even showing a higher IN selectivity. Conversely, the presence of the strong electron withdrawing unit NO₂, also able to act as an H-bond acceptor as in compound 23, was a more successful substitution for IN activity while still maintaining selectivity against RNase H function. The presence of the more aromatic thiophene ring led to the inactive compound 24. As already observed with 23, once again the introduction of a NO₂ moiety appeared related to the IN inhibitory activity, as compound 25 is the most active and selective of the five member ring series.

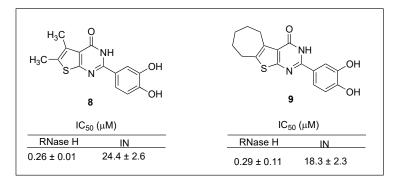


Figure 9. HIV-1 RT-RNase H and IN LEDGF/p75-dependent activity of compounds 8 and 9.

Interestingly, and conversely to what was previously observed for RNase H [30,31], the replacement of the dihydroxyphenyl core with other heteroaromatic nuclei had a greater impact on IN inhibitory activity.

As the indole moiety proved to be a key factor in some recently published IN allosteric inhibitors [43], we also examined compound **26**, whose activity against RT- RNase H has been reported [30]. Interestingly, it was most active against IN, although more selective for the RNase H function (Table 3), making it a good candidate for further optimization. Replacement of the indole moiety with pyrrole or 1-methyl pyrrole, as in compounds **27** and **28**, caused a loss of the IN activity and also affected RNAse H inhibition.

Finally, our attempt to investigate the mechanism of IN inhibition revealed the inability of the tested compounds to chelate Mg^{2+} ions. In fact, no Mg^{2+} effect was observed, as the absorbance peaks of our compounds did not show a significant shift with respect to control compound RDS1643, even at a high concentration of Mg^{2+} ions, suggesting a possible allosteric mode of action.

4. Materials and Methods

4.1. Chemistry

Unless otherwise specified, all reagents and solvents were obtained from commercial suppliers and used without further purification.

Progression of the reaction was monitored by thin-layer chromatography [Merck silica gel 60 pre-coated plates with fluorescent indicator F254 (0.25 mm)]. ¹H and ¹³C NMR spectra were recorded on a Bruker Avance IIITM HD 600 MHz spectrometer at 600 and 151 MHz, respectively. HPLC/APCI-MS analysis were recorded using a Varian tandem mass spectrometer (Palo Alto, CA, USA) 1200 L triple quadrupole mass spectrometer with an APCI source. Data acquisition and processing were performed by a Varian MS workstation version 6.7 software. APCI ran in both positive and negative ion modes. The capillary potential was set at 75 V, the APCI torch at 450 °C and the shield at 600 V. Nitrogen at 48 mTorr was fixed at 400 °C. The range of 100–1000 amu, with scan time 0.75 amu, scan width 0.70 amu, and detector at 1450 V were set to obtain the full scan spectra. The atmospheric pressure ionization (API) housing was kept at 50 °C for APCI. The compounds were solubilized in a mixture of acetonitrile 90% (v/v) (A) and doubly distilled water 10% (v/v) (B), and then infused in the source at a rate of 0.05 mL/min.

4.1.1. General Procedure for the Synthesis of Thienopyrimidinones *General procedure for synthesis of compounds* **10–12**, **19–22**, **24**, **25**, **27** *and* **28**.

Firstly, the 2-amino-4,5-disubstituted-thiophene-3-carboxamide intermediates were prepared as follows. A mixture of the appropriate ketone (20 mmol), cyanoacetamide (20 mmol), diethylamine (22 mmol, sulfur (20 mmol) and dry ethanol (20 mL) was stirred overnight at room temperature. After completion, the reaction was quenched with 140 mL of water and extracted five times with a mixture of ethyl acetate/ethanol (3/1). The organic layer, previously dried over Na₂SO₄, was filtered and the solvent was evaporated under vacuum to give a brown oil or solid. The crude product was then crystallized from ethanol, giving the corresponding intermediate. Then, the 2-amino-4,5-disubstituted-thiophene-3-carboxamide (1.8 mmol), molecular iodine (2.35 mmol), the proper aromatic aldehyde (2.2 mmol) and acetonitrile as solvent (25 mL) were stirred at room temperature for the period indicated (TLC) (1–5 h). After reaction, the mixture was quenched with 45 mL of 5% Na₂S₂O₃ solution, and the resulting precipitate was collected by filtration under vacuum. The crude product was washed with petroleum ether/ethyl acetate (1:1) and recrystallized from ethanol to give the final products 10–12, 19–22, 24, 25, 27 and 28. The purity of the compounds was assessed to be \geq 95% by HPLC analysis (Agilent HP-1100, Santa Clara, CA, USA).

4.1.2. Characterization Data

2-(2,3-*dihydroxyphenyl*)-5,6-*dimethylthieno*[2,3-*d*]*pyrimidin*-4(3H)-one (**10**). Yield (%) = 87. ¹H NMR (600 MHz, DMSO) δ 12.14 (bs, 1H), 7.60 (d, *J* = 8.1 Hz, 1H), 6.98 (dd, *J* = 13.3, 7.3 Hz, 1H), 6.78 (t, *J* = 7.9 Hz, 1H), 2.41 (s, 3H), 2.37 (s, 3H) ppm. ¹³C NMR (151 MHz, DMSO) δ 158.51, 147.35, 146.72, 130.02, 129.32, 121.97, 119.51, 118.85, 118.83, 115.90, 13.29, 13.09 ppm. HRMS calculated for C₁₄H₁₂N₂O₃S 288.0569 found 288.0570.

2-(2-hydroxyphenyl)-5,6-dimethylthieno[2,3-d]pyrimidin-4(3H)-one (**11**). Yield (%) = 90. ¹H NMR (600 MHz, DMSO) δ 12.14 (bs, 1H), 8.11 (d, *J* = 7.7 Hz, 1H), 7.42 (t, *J* = 7.8 Hz, 1H), 7.01 (d, *J* = 8.1 Hz, 1H), 6.98 (d, *J* = 7.6 Hz, 1H), 2.42 (s,3H), 2.39 (s, 3H) ppm. ¹³C NMR (151 MHz, DMSO) δ 160.84, 158.09, 153.26, 140.98, 133.63, 129.21, 119.95, 117.90, 111.57, 13.31, 13.12 ppm. HRMS calculated for C₁₄H₁₂N₂O₂S 272.0620 found 272.0621.

2-(2,4-*dihydroxyphenyl*)-5,6-*dimethylthieno*[2,3-*d*]*pyrimidin*-4(3*H*)-one (**12**). Yield (%) = 88. ¹H NMR (600 MHz, DMSO) δ 12.15 (s, 1H), 8.48 (s, 1H), 8.04 (d, *J* = 8.5 Hz, 1H), 7.47 (s, *J* = 8.5 Hz, 1H), 2.40 (s, 3H), 2.36 (s, 3H) ppm. ¹³C NMR (151 MHz, DMSO) δ 62.96, 162.61, 157.42, 147.02, 132.93, 132.15, 130.42, 127.63, 112.70, 103.69, 102.90, 13.30, 13.08. HRMS calculated for C₁₄H₁₂N₂O₃S 288.0620 found 288.0619.

2-(3,4-dihydroxyphenyl)-5-methyl-5,6,7,8-tetrahydrobenzo[4,5]thieno[2,3-d]pyrimidin-4(3H)-one (19). Yield (%) = 82. ¹H NMR (600 MHz, DMSO) δ 12.15 (bs, 1H), 9.63 (bs, 1H), 9.22 (bs, 1H), 7.58 (d, *J* = 2.2 Hz, 1H), 7.50 (dd, *J* = 8.7, 2.2 Hz, 1H), 6.81 (d, *J* = 8.7 Hz, 1H), 2.84–2.59 (m, 2H), 1.82 (s, 4H), 1.66 (d, *J* = 11.8 Hz, 1H), 1.25 (d, *J* = 6.9 Hz, 3H) ppm. ¹³C NMR (151 MHz, DMSO) δ 163.59, 160.32, 145.94, 145.83, 143.77, 137.71, 134.62, 134.12, 122.54, 121.29, 120.86, 115.44, 31.95, 30.36, 26.57, 25.80, 22.25 ppm. HRMS calculated for C₁₇H₁₆N₂O₃S 328.0882 found 328.0880.

2-(3,4-dihydroxyphenyl)-5-isobutylthieno[2,3-d]pyrimidin-4(3H)-one (**20**). Yield (%) = 80. ¹H NMR (600 MHz, DMSO) δ 12.33 (bs, 1H), 9.72 (bs, 1H), 9.29 (bs, 1H), 7.55 (d, *J* = 2.4 Hz, 1H), 7.50 (d, *J* = 8.4 Hz, 1H), 6.85–6.79 (m, 1H), 3.00 (d, *J* = 7.1 Hz, 2H), 2.09–2.01 (m, 1H), 0.93 (d, *J* = 6.5 Hz, 6H) ppm. ¹³C NMR (151 MHz, DMSO) δ 168.66, 159.34, 155.61, 150.41, 149.41, 143.77, 137.71, 134.62, 123.41, 119.99, 115.95, 115.84, 35.40, 27.66, 23.90 ppm. HRMS calculated for C₁₆H₁₆N₂O₃S 316.0882 found 316.0881.

2-(*furan*-2-*y*])-5,6-*dimethylthieno*[2,3-*d*]*pyrimidin*-4(3*H*)-one (**21**). Yield (%) = 92. ¹H NMR (600 MHz, DMSO) δ 12.46 (s, 1H), 7.97 (d, *J* = 3.5 Hz, 1H), 7.58 (d, *J* = 3.5 Hz, 1H), 6.76–6.71 (m, 1H), 2.41 (s, 3H), 2.37 (s, 3H) ppm. ¹³C NMR (151 MHz, DMSO) δ 162.53, 158.94, 156.49,

149.47, 144.38, 135.38, 129.37, 121.80, 116.11, 113.10, 109.52, 103.74, 13.32, 13.10 ppm. HRMS calculated for $C_{12}H_{10}N_2O_2S$ 260.0463 found 260.0466.

5,6-dimethyl-2-(5-methylfuran-2-yl)thieno[2,3-d]pyrimidin-4(3H)-one (**22**). Yield (%) = 91. ¹H NMR (600 MHz, DMSO) δ 12.33, 7.50, 7.50, 6.36, 6.36, 3.41, 2.52, 2.51, 2.51, 2.51, 2.50, 2.40, 2.39, 2.36 ppm. ¹³C NMR (151 MHz, DMSO) δ 162.53, 158.94, 156.49, 144.38, 144.14, 129.58, 129.37, 121.80, 116.11, 109.52, 13.99, 13.32, 13.10 ppm. HRMS calculated for C₁₃H₁₂N₂O₂S 260.0619 found 260.0621.

5,6-dimethyl-2-(thiophen-2-yl)thieno[2,3-d]pyrimidin-4(3H)-one (24). Yield (%) = 86. ¹H NMR (600 MHz, DMSO) δ 12.61 (bs, 1H), 8.21 (d, *J* = 3.8 Hz, 1H), 7.85 (d, *J* = 5.1 Hz, 1H), 7.22 (ddd, *J* = 5.1, 3.8, 1.5 Hz, 1H), 2.41 (s, 3H), 2.37 (s, 3H) ppm. ¹³C NMR (151 MHz, DMSO) δ 155.26, 148.04, 147.97, 132.09, 132.03, 130.53, 130.18, 125.02, 122.61, 117.66,13.38, 13.27 ppm. HRMS calculated for C₁₇H₁₆N₂OS₃ 360.0425 found 360.0428.

5,6-dimethyl-2-(5-nitrothiophen-2-yl)thieno[2,3-d]pyrimidin-4(3H)-one (**25**). Yield (%) = 93. ¹H NMR (600 MHz, DMSO) δ 7.84 (d, J = 4.0 Hz, 1H), 7.80 (d, J = 4.1 Hz, 1H), 2.41 (s, 3H), 2.39 (s, 3H) ppm. ¹³C NMR (151 MHz, DMSO) δ¹³C NMR (151 MHz, DMSO) δ 169.86, 163.66, 150.00, 143.74, 133.73, 131.06, 111.07, 103.24, 13.30, 13.10 ppm. HRMS calculated for C₁₂H₉N₃O₃S₂ 307.0085 found 307.0088.

5,6-dimethyl-2-(1H-pyrrol-2-yl)thieno[2,3-d]pyrimidin-4(3H)-one (**27**). Yield (%) = 84. ¹H NMR (600 MHz, DMSO) δ 12.13 (bs, 1H), 11.78 (bs, 1H), 7.28 (d, *J* = 3.8 Hz, 1H), 7.00 (d, *J* = 3.4 Hz, 1H), 6.20 (d, *J* = 3.4 Hz, 1H), 2.47–2.29 (m, 6H) ppm. ¹³C NMR (151 MHz, DMSO) δ 162.54, 158.95, 147.97, 134.24, 133.80, 129.26, 129.13, 125.28, 125.02, 13.11, 12.14 ppm. HRMS calculated for C₁₇H₁₇N₃OS₂ 343.0813 found 343.0816.

5,6-*dimethyl*-2-(1-*methyl*-1H-*pyrrol*-2-*yl*)*thieno*[2,3-*d*]*pyrimidin*-4(3H)-*one* (**28**). Yield (%) = 81. ¹H NMR (600 MHz, DMSO) δ 12.40 (bs, 1H), 7.16 (d, *J* = 13.2 Hz, 1H), 7.06 (d, *J* = 13.2 Hz, 1H), 6.13 (d, *J* = 3.2 Hz, 1H), 3.62 (s, 3H), 2.50 (s, 3H), 2.36 (s, 3H) ppm. ¹³C NMR (151 MHz, DMSO) δ 166.82, 152.52, 145.94, 145.83, 134.62, 134.12, 120.54, 119.99, 118.83, 115.84, 41.13, 12.07 ppm. HRMS calculated for C₁₈H₁₉N₃OS₂ 357.0970 found 357.0973.

4.2. Biology

4.2.1. Expression and Purification of HIV-1 Integrase and LEDGF/p75 Proteins

Recombinant His and FLAG-IN proteins and His and FLAG-LEDGF proteins were expressed in *Escherichia coli* strain BL21 (DE3) and purified, as previously described [44].

4.2.2. HTRF LEDGF-Dependent Assays

IN activity in the presence of recombinant LEDGF/p75 protein were determined as described before [45]. IN was pre-incubated in reaction buffer containing 20 mM HEPES pH 7.5, 1 mM DTT, 1% glycerol, 20 mM MgCl₂, 0.05% Brij-35 and 0.1 mg/mL BSA with increasing concentration of compounds for 1 h at room temperature. After the addition to this mixture of 9 nM DNA donor substrate, 50 nM DNA acceptor substrate and 50 nM LEDGF/p75 protein (when required), the reaction was incubated at 37 °C for 90 min. The addition of 4 nM of Europium–Streptavidin to the reaction mixture provided the HTRF signal, which was recorded using a Perkin Elmer (Waltham, MA, USA) Victor 3 plate reader with 314 nm for excitation wavelength and 668 and 620 nm for the wavelength of the acceptor and the donor substrates emission, respectively.

4.2.3. MgCl₂ Coordination Assay

The ability of the compounds to coordinate Mg^{2+} were determined by means of previously reported protocols [31]. The UV–Vis spectrum from 250 to 600 nm was recorded before and after the addition of 6 mM MgCl₂ and plotted using Prism9.

4.2.4. RNase H inibitory Assay

The previously reported procedure was used for preparation and purification of wild type p66/p51 HIV-1 RT. Enzyme was stored in a 50% glycerol-containing buffer at

-20 °C [32,46]. An 18-nt 3'-fluorescein-labeled RNA annealed to a complementary 18-nt 5'dabsyl-labeled DNA was used to determine IC₅₀ values, following a previously described procedure [47]. In brief, 1 µL of each inhibitor dissolved in DMSO was added to a 96- well plate, followed by 10 µL of the proper RT (15–80 ng/mL) in reaction buffer. The addition of 10 µL of RNA/DNA hybrid (2.5 µM) initiated the hydrolysis. Then, 50 mM Tris-HCl, pH 8.0, 60 mM KCl, 10 mM MgCl₂, 1% DMSO, 150–800 ng of RT, 250 nM substrate, and increasing concentrations of inhibitor were chosen as final assay conditions. As negative control, wells containing only DMSO were used. Subsequently, the plates were incubated in a Spectramax Gemini EM fluorescence spectrometer for 10 min at 37 °C, measuring the fluorescence (λ ex = 485 nm; λ em = 520 nm) at 1 min intervals in order to determine the linear initial rates in the presence (vi) and absence (vo) of inhibitor. Percent inhibition was assessed as 100(vo – vi)/vo and plotted against log[I]. Prism5 (GraphPad Software 5.0) was used to calculate the IC₅₀ values. All assays were carried out in triplicate.

5. Conclusions

Since the dihydroxyphenyl unit has been evidenced in various allosteric IN inhibitors, we investigated here the activity of novel thienopyrimidinones bearing a dihydrohyphenyl core at C-2. Our findings showed a strict correlation between the OH position and their IN inhibitory activity, and coordination assays evidenced no Mg²⁺ effect, suggesting an allosteric mode of action.

In conclusion, the thienopyrimidinone skeleton is confirmed as a privileged framework in HIV drug discovery. Furthermore, as catechol may be implicated in cytotoxicity, we also explored its substitution with other heteroaromatic scaffolds. In particular, our findings indicated that variation at C-2 position may be a key factor for compound selectivity. In this regard, compounds **21–23** and **25–26** with an improved activity were revealed to be a good hit to lead candidates. In particular, compound **26** showed very good activity on both RNase H and IN and is proposed as a dual-inhibitor, while 5,6-dimethyl-2-(5-nitrothiophen-2-yl)thieno[2,3-d]pyrimidin-4(3H)-one **25**, which was completely inactive on RNase H but with a good IN IC₅₀, was the most selective against IN.

Substitutions at C-5 and C-6 also proved to be vital. In this regard, the crucial introduction of a branched cyclohexyl moiety instead of a cycloheptane ring offered the 2-(3,4-dihydroxyphenyl)-5-methyl-5,6,7,8-tetrahydrobenzo[4,5]thieno[2,3-d]pyrimidin-4(3H)-one **20** as a hit compound for further exploitation. Encouraged by the results, novel derivatives are now under consideration, also using an in silico approach, to gain more insight into the allosteric binding site (manuscript in preparation).

Supplementary Materials: The following supporting information can be downloaded at: https: //www.mdpi.com/article/10.3390/molecules28186700/s1, ¹H and ¹³C NMR spectra: pp. 2–23. Mg²⁺ spectra: pp. 24–32.

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